CHEMV12B, Organic Chemistry II

1.) Name three structural differences between DNA and RNA. (12 pts, 4 pts ea)

(1) DNA: double helix (2 strands) vs. RNA: single strand

(2) DNA: thymine vs. RNA: uracil

(3) DNA: deoxyribose vs. RNA: D-ribose

2.) For each pair of amino acids, (1) identify each by name, (2) classify each as acidic/polar, basic/polar, aromatic, aliphatic, neutral polar, sulfur, and/or twisty, and (3) list all possible interactions between the two that would hold together its tertiary structure. (80 pts, 20 pts ea)

a.)

(1) histidine

(1) glutamic acid

(2) basic/polar, aromatic

(2) acidic/polar

(3) salt bridges (acid + base), hydrogen bonding (both polar)

b.)

(1) asparagine

(1) serine

(2) neutral polar

(2) neutral polar

(3) hydrogen bonding (both polar), metallic ion coordination (both partially negative)

$$H_3N^+$$
 and H_3N^+ O

c.)

(1) glycine

(1) valine

(2) twisty

(2) aliphatic

(3) hydrophobic interactions (both nonpolar)

$$H_3N^+$$
 o and NH_3^+

d.)

(1) phenylalanine

(1) isoleucine

(2) aromatic

(2) aliphatic

(3) hydrophobic interactions

3.) Consider the following fragment of a nucleic acid. (16 pts)

3'-GCATACGCCGCAACTATT...

a.) Is this DNA or RNA? Explain.

DNA - contains thymine

b.) Give the sequence of the *complementary* strand's matching fragment.

5'-CGTATGCGGCGTTGATAA...

c.) Give the sequence of the product after *transcription* of the leading strand's fragment.

5'-CGUAUGCGGCGUUGAUAA

d.) Give the sequence of the product after translation of this fragment.

Met-Arg-Arg

- 4.) Consider a generic dipeptide. (30 pts)
 - a.) Show the general mechanism for its formation. Show both amino acids in their *zwitterionic* forms. Indicate the side chains on each amino acid with "R". Show all *chirality* (wedges and dashes) and *geometry*.

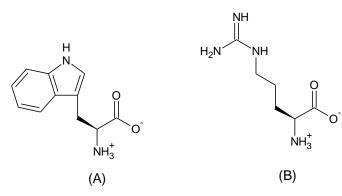
$$H_3N^+$$
 H_3N^+
 H_3N^+
 H_3N^+
 H_3N^+
 H_3N^+
 H_3N^+
 H_2N^+
 H

- b.) In the product above, identify the (1) C-terminus, (2) N-terminus, (3) amide linker and (4) peptide bond.
- c.) Classify this type of reaction in two ways.
 - (1) condensation
- (2) dehydration
- 5.) Identify the type of molecule that forms the following secondary structures. (12 pts, 3 pts ea)

a.) α-helix	<u>proteins</u>		
b.) double helix	DNA		
c.) single strand knots	RNA		
d.) β-sheet	proteins		

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6.) Consider the following amino acids. (25 pts)



a.) Name (A).

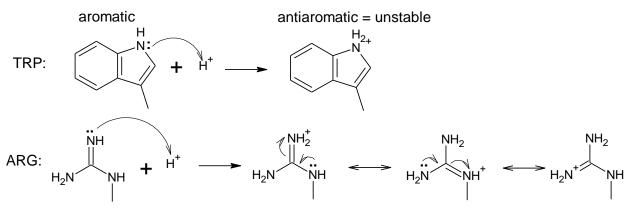
tryptophan

b.) Name (B).

arginine

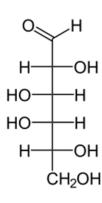
c.) Which amino acid is overall *more* basic? **Explain**. Your answer should include any relevant resonance

structures. You may abbreviate the structures to show only the relevant sections.



delocalized = stable, VERY basic

7.) Consider the following carbohydrate. (25 pts)



- b.) Give the common structure name for this carbohydrate.

aldohexose

c.) Is this a reducing sugar? **Explain**.

YES - contains CHO group that can be oxidized to COOH

d.) Draw the Haworth projection for this carbohydrate in α –cyclic form. (1) Star the anomeric carbon and (2) circle the hemiacetal functional group.