

# Common Conversion Factors

## Chem20, Elementary Chemistry

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### Length (base unit)

*SI unit: meter (m)*

$$1 \text{ mile} = 5280 \text{ feet}$$
$$= 1.61 \text{ km}$$

$$1 \text{ m} = 39.37 \text{ inches}$$

$$1 \text{ in.} = 2.54 \text{ cm} \text{ (definition)}$$

$$1 \text{ \AA} = 10^{-10} \text{ m}$$

### Mass (base unit)

*SI unit: kilogram (kg)*

$$1 \text{ kg} = 2.205 \text{ pounds (lb)}$$

$$1 \text{ lb.} = 453.6 \text{ g}$$
$$= 16 \text{ oz.}$$

$$1 \text{ ton} = 2000 \text{ lb.}$$

$$1 \text{ metric ton} = 1000 \text{ kg} = 1.103 \text{ tons}$$

$$1 \text{ g} = 6.022 \times 10^{23} \text{ atomic mass units (amu)}$$

### Temperature (base unit)

*SI unit: Kelvin (K)*

$$K = ^\circ C + 273.15$$

$$^\circ C = 5/9(^\circ F - 32)$$

$$^\circ F = 9/5(^\circ C) + 32$$

### Energy (derived unit)

*SI unit: Joule (J)*

$$1 \text{ J} = 1 \text{ N} \cdot \text{m}$$

$$1 \text{ calorie} = 4.184 \text{ J} \text{ (definition)}$$

$$1 \text{ L} \cdot \text{atm} = 101.33 \text{ J}$$

### Pressure (derived)

*SI unit: Pascal (Pa)*

$$1 \text{ Pa} = 1 \text{ N/m}^2$$

$$1 \text{ atm} = 101325 \text{ Pa} \text{ (definition)}$$

$$= 1.01325 \text{ bar}$$

$$= 760 \text{ torr} \text{ (definition)}$$

$$= 14.7 \text{ lb/in.}^2$$

$$1 \text{ torr} = 1 \text{ mmHg}$$

### Volume (derived)

*SI unit: cubic meter (m<sup>3</sup>)*

$$1 \text{ L} = 10^{-3} \text{ m}^3$$

$$= 10^3 \text{ cm}^3$$

$$= 1.0567 \text{ qt.}$$

$$1 \text{ gallon} = 4 \text{ quarts}$$

$$= 8 \text{ pints}$$

$$= 128 \text{ fluid ounces}$$

$$= 3.7854 \text{ L}$$

$$1 \text{ cm}^3 = 1 \text{ mL} \text{ (definition)}$$